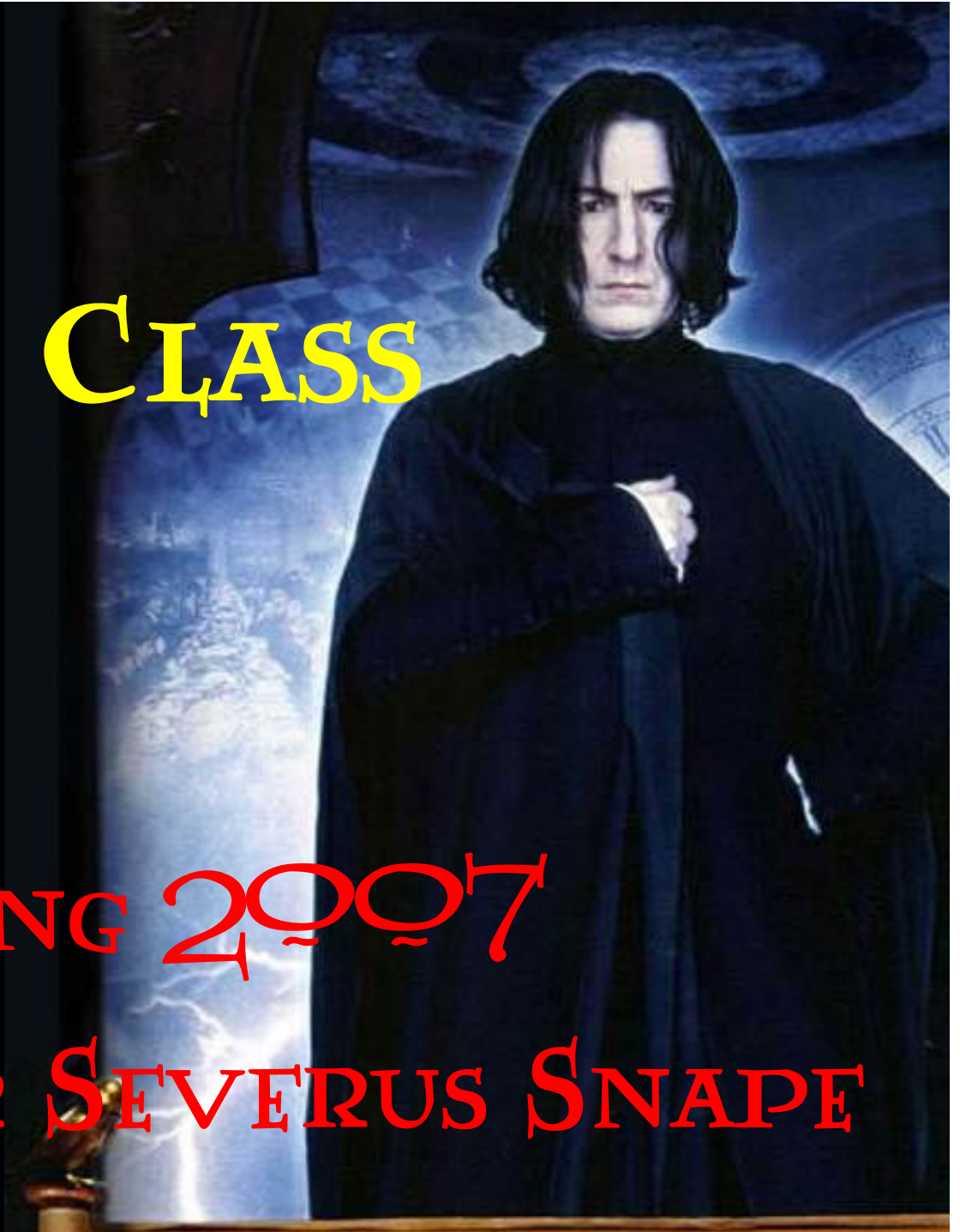


CH204

POTIONS CLASS

SPRING 2007

PROFESSOR SEVERUS SNAPE



LAST WEEK IN THE POTIONS LABORATORY

SEPARATED MIXTURES BASED ON DIFFERING
PHYSICAL AND CHEMICAL PROPERTIES

USED EXCEL TO CALCULATE AVERAGE, STANDARD
DEVIATION, AND WEIGHT PERCENTS

Q-TEST IN ACTION!

BUCHNER FILTERING

EXPERIMENT 3
QUALITATIVE CHEMICAL ANALYSIS

“QUANT” VS “QUAL”

QUANTITATIVE – HOW MUCH IS THERE?

QUALITATIVE – WHAT IS IT?

YOU WILL IDENTIFY THE CHEMICAL
IDENTITIES OF 5 UNKNOWN SOLUTIONS
BASED ON HOW THEY REACT
(OR DON'T REACT!)
WITH ONE ANOTHER.

PREVIOUS YEARS

GORGON'S BLOOD

LIQUID GOLD

VITREOUS HUMOR OF A BLIND MULE

2007

ACIDS

BASES

INORGANIC SALTS

TWO-PART LAB

- PART 1: MIX TEN KNOWN SOLUTIONS AND RECORD THE RESULTS OF THE REACTIONS
- PART 2: MIX YOUR FIVE UNKNOWNNS AND COMPARE THE RESULTS WITH WHAT YOU SAW IN PART ONE.

BE EXACT!

**THE MORE ACCURATELY YOU RECORD
YOUR OBSERVATIONS, THE EASIER IT
WILL BE TO IDENTIFY YOUR
UNKNOWNNS.**

WHAT ARE WE LOOKING FOR?

*EXPLOSIONS

*SUPERNATURAL CREATURES

*RASHES, MUTATIONS,

TRANSFORMATIONS



WHAT ARE WE ^{REALLY} LOOKING FOR?

PRECIPITATES.

(SEE THE SOLUBILITY TABLE IN APPENDIX 2.)

DON'T EXPECT TO SEE ANY ACID-BASE ACTION.

WRITE CHEMICAL EQUATIONS FOR ALL OF THE REACTIONS THAT FORM A PRECIPITATE.

THE KNOWN SOLUTIONS

ACIDS: HCl H₂SO₄ HNO₃

BASES: NaOH Na₂S

SALTS: Ba(NO₃)₂ AgNO₃ K₂CrO₄

Fe(NO₃)₃ Ni(NO₃)₂

ALL SOLUTIONS ARE 0.10 OR 0.20 M.

RIDDLE ME THIS

WHAT DO YOU GET WHEN YOU CROSS
HYDROCHLORIC ACID WITH SILVER NITRATE?

BALANCED CHEMICAL EQUATION



ADD THE PHYSICAL STATES OF EACH COMPOUND

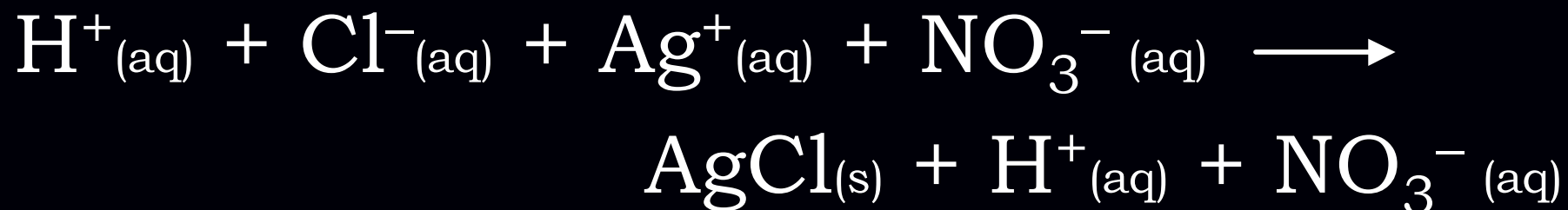


THIS IS CALLED A MOLECULAR EQUATION.

LET'S GET REAL



WRITE AQUEOUS COMPOUNDS AS INDIVIDUAL IONS:

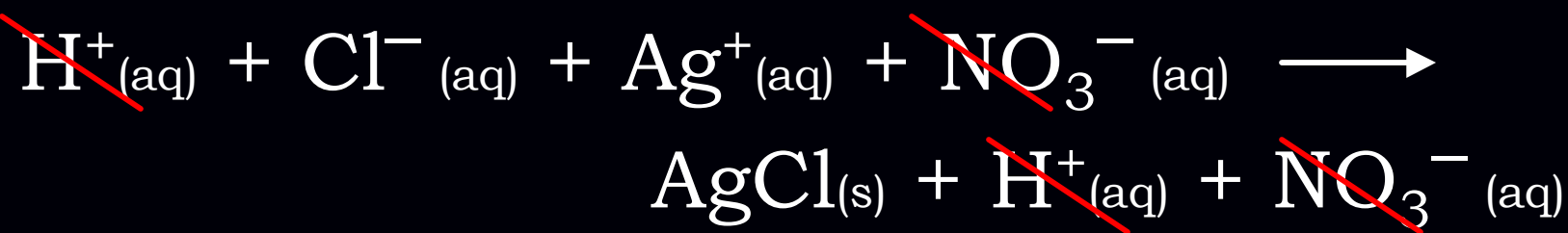


THIS IS A TOTAL IONIC EQUATION.

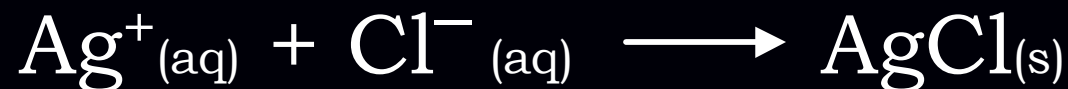
LOTS OF SPECTATOR IONS.

TIME TO CLEAN HOUSE

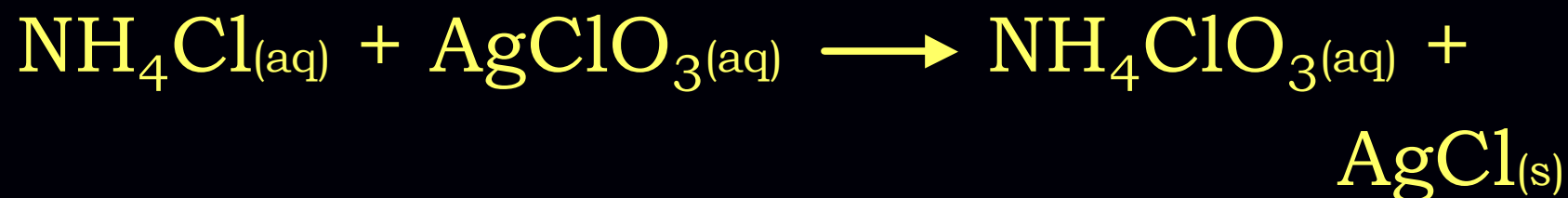
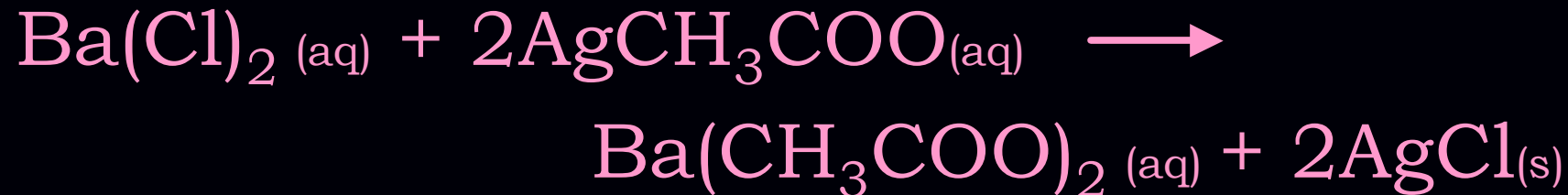
CROSS OUT SPECTATOR IONS



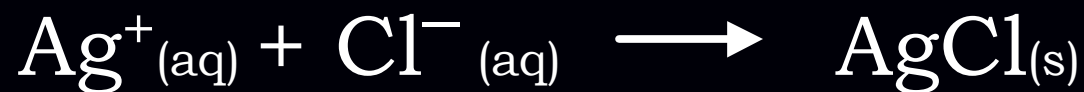
THIS LEAVES US WITH A NET IONIC EQUATION



THE NET IONIC EQUATION



ALL OF THESE REACTIONS HAVE THE SAME NET IONIC EQUATION:



SIMPLE IS GOOD

- THE NET IONIC EQUATION DESCRIBES THE CHEMICAL REACTION THAT OCCURS, AND DOES NOT INCLUDE ANY IONS THAT DO NOT TAKE PART IN THE REACTION, *EVEN THOUGH THOSE IONS ARE PRESENT IN SOLUTION.*
- HOW DO WE KNOW WHICH IONS WILL REACT AND WHICH ONES WON'T?

SOME QUICK SOLUBILITY RULES

★ ALL COMPOUNDS CONTAINING ALKALI METALS AND AMMONIUM ION ARE SOLUBLE.



★ ALL COMPOUNDS CONTAINING NITRATE, CHLORATE, PERCHLORATE, AND ACETATE ARE SOLUBLE.



SOME QUICK INSOLUBILITY RULES

- ★ ALL COMPOUNDS CONTAINING PO_4^{3-} , CO_3^{2-} , AND SO_3^{2-} IONS ARE INSOLUBLE EXCEPT THOSE THAT ALSO CONTAIN ALKALI METALS OR NH_4^+ .
- ★ ALL HYDROXIDES (OH^-) AND SULFIDES (S^{2-}) ARE INSOLUBLE EXCEPT THOSE THAT ALSO CONTAIN ALKALI METALS, OR NH_4^+ . (EXCEPTION: SOME GROUP II HYDROXIDES ARE SLIGHTLY SOLUBLE.)
- ★ WHEN IN DOUBT, Ag^+ Pb^{2+} AND Hg COMPOUNDS TEND TO BE INSOLUBLE.

IN THE POTIONS LABORATORY

★ CREATE AN ARRAY OF REACTIONS IN THE MICROWELL PLATE SIMILAR TO THE ONE IN THE LAB MANUAL.

★ USE ONLY 2 DROPS OF EACH REACTANT.

★ *DO NOT* TOUCH THE TIPS OF THE DROPPER BOTTLES TO THE SOLUTIONS IN THE MICROWELL PLATE OR YOU WILL DIE A MOST PAINFUL DEATH.

VILE, HIDEOUS FLUIDS!

EMPTY YOUR USED MICROWELL PLATES
INTO THE DISGUSTING PLASTIC TRAY IN THE HOOD.

RINSE THE PLATES WITH ORDINARY TAP WATER,
DO A FINAL RINSE WITH DEION WATER,
AND STACK THE PLATES IN THE HOOD.

LAB REPORT

MOLECULAR EQUATIONS FOR
11 PRECIPITATION REACTIONS.

NET IONIC EQUATIONS FOR
11 PRECIPITATION REACTIONS.

11 + 11 is 22 EQUATIONS ALTOGETHER.

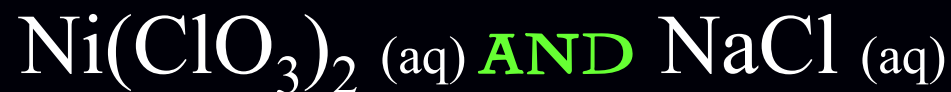
WARNING! EVIL MAGIC!

SOME REACTIONS WILL TURN CLOUDY
EVEN THOUGH NO SOLID IS FORMED.



ALL PRECIPITATES WILL BE INSOLUBLE IN THE
SOLUBILITY TABLE. DO NOT COUNT ANY
“SLIGHTLY SOLUBLES” AS PRECIPITATES.

POST-LAB PREVIEW



WHAT WILL PRECIPITATE WITH Ni^{2+}

BUT NOT WITH Na^+ ?

OR

WHAT WILL PRECIPITATE WITH Cl^-

BUT NOT WITH ClO_3^- ?

QUESTION 1D – WE DID THIS IN LAB LAST WEEK.

NEXT WEEK

EXPERIMENT 4: ACID-BASE TITRATION

- MORE INVOLVED THAN THIS LAB.
- REQUIRES MORE PREPARATION THAN OUR PREVIOUS LABS.
- **START PREPARING EARLY**

PRE-LAB QUESTION 1: THE ANSWER IS NOT 71!

QUIZ TIME

- * YOU WILL NEED A CALCULATOR EVERY WEEK (EXCEPT NEXT WEEK).
- * MAKE SURE YOU KNOW YOUR SECTION NUMBER AND YOUR TA'S NAME.

